

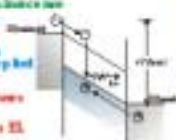
## I. Introduction

**EL: Electro-luminescence**  
The light comes from the recombination of electrons and holes in an applied electric field.  
Depend on the materials used for the active materials in organic EL.

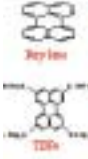
### OLED Material design and development

For Dr. H. Y. Tang, Prof. S. T. Lee, Department of Physical and Materials Science

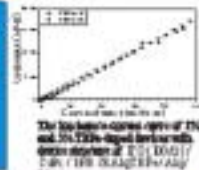
- There are several functions in OLED: hole-transport material, electron-transport material, emitting material (include host material and dopant material)
- Developing of efficient OLED materials are aimed at improving the color purity, energy efficiency and operation stability.



## II. Better Blue OLED Dopant Material by Reduction of Aggregation (TRPs)



TRPs (Tetra-Resonance Polymers) are a class of materials designed to reduce the aggregation of dopant molecules in OLEDs. They are used as dopant materials in blue OLEDs to improve the efficiency and stability of the devices.



## III. Pure color and high efficient green dopant material (PNA)



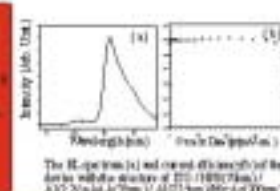
PNA (Pure color and high efficient green dopant material) is a class of materials designed to improve the efficiency and stability of green OLEDs. It is used as a dopant material in green OLEDs.

The device structure is shown in the figure. The device is a green OLED with a structure of ITO/HTL/PNA/ETL/Cathode. The current density of the device is shown in the graph.

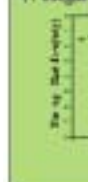
## IV. Red dopant material (DyNA, pAAA)



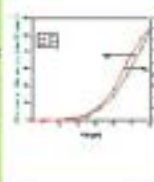
The absorption and emission spectra of the red dopant material are shown in the figure. The absorption peak is at 410 nm and the emission peak is at 620 nm. The device structure is shown in the figure.



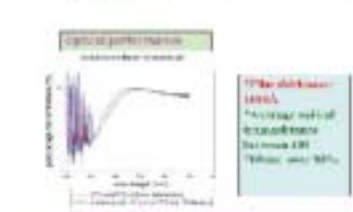
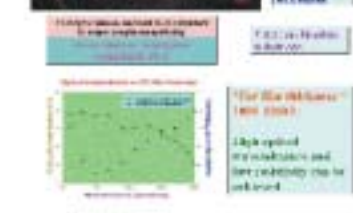
## V. High Stable Hole Transport Material (HPCz)



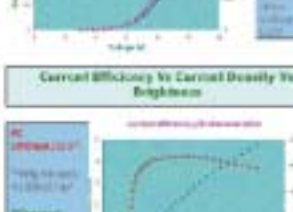
HPCz (High Stable Hole Transport Material) is a class of materials designed to improve the efficiency and stability of OLEDs. It is used as a hole transport material in OLEDs.



## Low Temperature Deposition of Transparent Conductive Oxide (TCO) on plastic substrate and fabrication of Flexible Organic Light emitting Diode



## Configuration of PLEDs with novel hole TCO



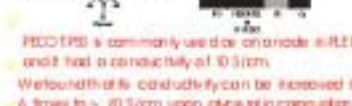
## Device fabrication and characterization of Red emitting OLEDs

Device fabrication and characterization of Red emitting OLEDs.

### I. Anode Modification in REDs

Ultraviolet Photoemission Spectroscopy (UPS) revealed that the low work function of Ca led to no electron injection barrier at the FI (organic emitting polymer) interface.

Device efficiency in REDs is basically limited by the hole injection.



PFEDOPB is commonly used as an anode in REDs and had a conductivity of 10<sup>3</sup> S/cm. Without the hole conductivity can be increased by 4 times to ~40 S/cm upon glycerol incorporation.

The incorporation of glycerol into the PFEDOPB (G-PFEDOPB) has no observable side effect on the device electro-structure.

PFEDs with the configuration of FTO/PEDOT/PVCA have fabricated since to reduce the performance of the G-PFEDOPB buffer layer.

### II. Cathode Modification in SM-LEDs

Mg:Ag is often used as a cathode in SM-LEDs. Practical applications using OLEDs are feasible only if the devices have sufficiently long life time.

Reaction between cathode and water/oxygen species, i.e. cathode oxidation is one of the causes for formation and growth of dark spots.

Effective protection against oxygen and moisture is necessary here, a simple method to inhibit the growth of dark spots is prepared by deposition of a metal with higher E.C.P. on the cathode, e.g. Ca.



Since Ca has a more negative E.C.P. (-2.87 eV) than Mg (-3.4 eV), Ca is preferentially oxidized when galvanically couples to the Mg.

Ca is a sacrificial metal that oxidation was accelerated when Ca was completely consumed.

The delay of growth of dark spot is inhibited when other metals are used instead of Ca. Cu and In cathode is less effective and the protective function of Ca involved in redox reaction is more significant.



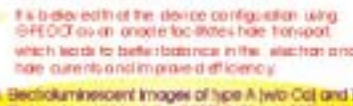
At a driving voltage of 6 V, the current density in the glycerol-doped device (3.8 mA/cm<sup>2</sup>) is double that in the undoped device (1.9 mA/cm<sup>2</sup>). The brightness of the doped device (1500 Cd/m<sup>2</sup>) is more than 100% higher than that of the undoped device (800 Cd/m<sup>2</sup>).

At a current density of 20 mA/cm<sup>2</sup>, the doped device shows much higher power (1.2 mW) and current (1.8 C/sA) efficiencies than those (0.8 mW and 1.3 C/sA) respectively of the undoped device.

The inset of left figure shows the resistance of the FTO/PEDOT/PVCA (2.7 MΩ) is about 5 times lower than that of the PFEDOPB/PVCA. These results suggest that adding glycerol can increase the conductivity of the PFEDOPB layer and the corresponding hole current in the device.

If a buffer layer of the device configuration using G-PFEDOPB as an anode for SM-LEDs has been reported, which leads to better balance in the electron and hole currents and improved efficiency.

### III. Backluminescent Images of type A (w/o Ca) and B (w/Ca) devices of luminance of 95% (left) and 60% (right)



Backluminescent images of type A (w/o Ca) and B (w/Ca) devices of luminance of 95% (left) and 60% (right).



Dark spots growth rate is  $d^2/dt^2$  in a diffusion-controlled process.

Devices with Ca can retard the growth rate of dark spots.

Since Ca has a more negative E.C.P. (-2.87 eV) than Mg (-3.4 eV), Ca is preferentially oxidized when galvanically couples to the Mg.

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The delay of growth of dark spot is inhibited when other metals are used instead of Ca. Cu and In cathode is less effective and the protective function of Ca involved in redox reaction is more significant.

## Surface and Interface Studies on Organic Electrochromic Devices by High Resolution Electron Energy Loss Spectroscopy

High Resolution Electron Energy Loss Spectroscopy (HREELS) studies on Organic Electrochromic Devices.



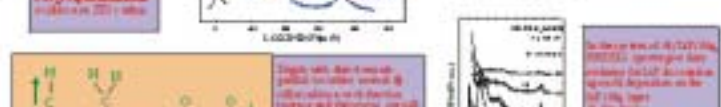
HREELS (High Resolution Electron Energy Loss Spectroscopy) studies on Organic Electrochromic Devices. The spectra show the characteristic peaks of the device.



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